

HER

Application of data

HER

ISO Requirements and Purser Furnace Characteristics

HER

**Fire Classification-BS7899-2(1999)
and ISO TR9122-4 (1993)**

HER

Yields in Flaming fires

HER

Toxic product yields in fires

Depend upon three major parameters:

HER

Main variables determining toxic product yields for any particular material

- Flaming or non-flaming decomposition
- For flaming decomposition the fuel/air (equivalence ratio)
- To a lesser extent, the oxygen concentration and compartment temperature
- Fuel composition

HER

The Equivalence Ratio ϕ

For well-ventilated fires, $\phi < 1$,

For fuel-rich (vitiated) combustion, $\phi > 1$

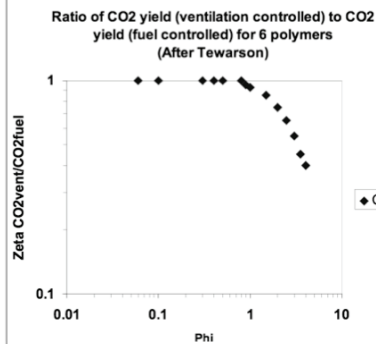
Further factors affecting yields of CO and other products:

- Oxygen concentration
- Temperature
- Fire retardants

HER

CO₂ yield and equivalence ratio

Effect of ϕ on CO₂ yield

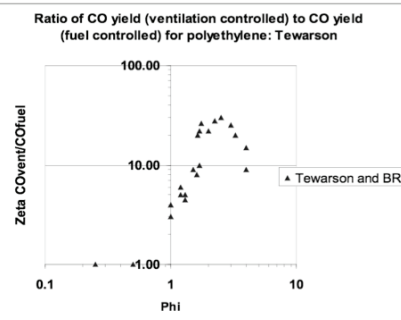


- Under well ventilated conditions almost all fuel carbon is oxidised to CO₂
- Above $\phi=1$ the yield decreases
- Data from Tewarson using ASTM E2058 fire propagation apparatus

HER

CO yield and equivalence ratio

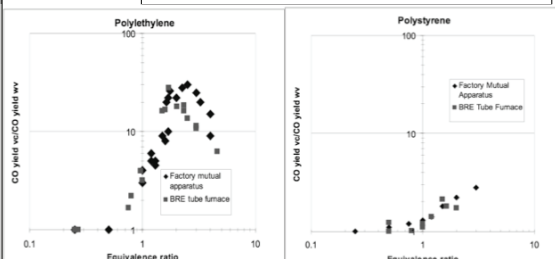
Effect of ϕ on CO yield



- CO yield is low under well-ventilated conditions
- Above $\phi=1$ the yield increases steeply
- Data from Tewarson using ASTM E2058 fire propagation apparatus

HER

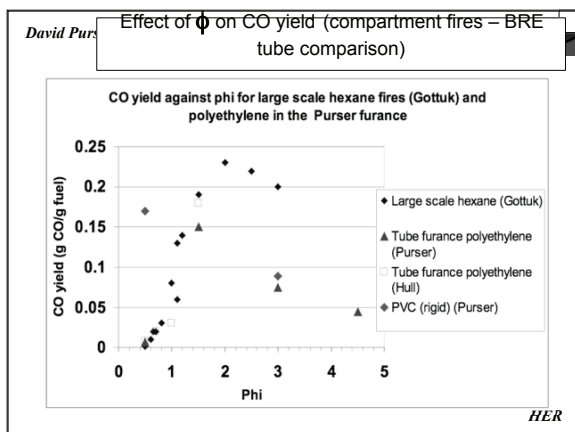
Effect of ϕ on CO yield (Tewarson – BRE comparison)



In early tube furnace experiments we found a good agreement between the ASTM E2058 and ISO TS19700 results expressed as a function of ϕ confirming that a fundamental (apparatus independent) property was being measured

Effect of ϕ on CO yield (Tewarson, compartment fires BRE tube furnace comparison)

HER



David Purser

Relationship between phi and product yields for 15 materials and large-scale validation for flaming fires

Early studies have demonstrated that for flaming fires the tube furnace:

- Measures the relationship between phi and product yields
- Provides very similar results to the Factory Mutual and BRI apparatus in relation to phi – suggesting that the results are apparatus independent and measuring relatively fundamental phenomena
- Gives a reasonably good agreement with large and full-scale fires – providing some validation of the method against full-scale data

Detailed study on a range of 15 common polymeric materials:

- Relationship with phi for flaming conditions
- Effects of temperature, oxygen concentration and fire retardants
- Validation against large scale fires for selected materials

HER

David Purser

Derived and measured parameters

- Mass charge concentration (mg/l)
- Mass loss concentration (mg/l)
- Effective heat of combustion (MJ/kg)
- Combustion efficiency
- phi
- CO₂/CO ratio (Volume ratio)
- Oxygen consumed (g/g)
- Mass loss / oxygen ratio (g/g)
- Smoke yield (SEA) (g/g)
- CO, CO₂, HCN, HCl, HBr, NO, NO_x, SO₂, PO₄ yields (g/g)
- Acrolein, formaldehyde yield individual GC-MS organics (g/g)
- Total carbon as organic carbon (using secondary furnace) (g/g)

HER

David Purser

Tube furnace combustion conditions

- Furnace temperature 650°C (700°C for PIR and MDF-FR) some additional experiments at 850°C)
- Constant flaming combustion**
- Most runs in air, but some in 10% or 12% oxygen using air/nitrogen mixtures

HER

David Purser

Equivalence ratio measurements

- Nominal equivalence ratio from stoichiometric oxygen demand for complete combustion of fuel: 0.5-2.5 in separate runs
- Actual equivalence ratio achieved calculated from total oxygen consumed in the secondary furnace (phi meter) plus that required for oxidation of soot.

$\phi_{actual} = \frac{\phi_{nominal} + \phi_{soot}}{\phi_{nominal}}$

For materials undergoing complete combustion nominal and actual phi very similar, for char formers the actual phi was lower than nominal due to formation of carbon-rich char (close to nominal values assuming char 100% carbon)

HER

David Purser

Large-scale validation experiments

- ISO 9705 room: MDF board or PIR aluminium-faced panels, and cribs of the same material used as radiation source in corner designed to give standard HRR (equivalent to propane flame)
- Half-scale ISO room corridor rig: MDF or PIR as cribs
- Conditions varied from well-ventilated to vitiated by varying door panel opening to obtain different global equivalence ratios
- Aimed to obtain a period of near steady-state burning during which effluent samples were taken from the upper layer for composition and equivalence ratio measurements
- For ISO room samples taken near ceiling before doorway, for half-scale rig near ceiling at back of room away from doorway (and in corridor).

HER

Two ways to report and compare yields

- Mass yield = mass of product (e.g. CO)/fuel mass loss (g/g)
- Mass recovery fraction = mass yield normalized in terms of carbon content (or nitrogen content) of fuel
- For CO this would be the mass of carbon released as CO as a fraction of the total carbon mass loss
- Mass yield = recovery fraction x carbon content of fuel mass loss (g)
- Recovery fractions therefore represent normalised yields showing the extent to which conversion to CO is independent of polymer composition or not

HER

Materials tested in the tube furnace

	% Fr components		
PMMA			
Polyethylene (low density)			
Polystyrene			
Polyamide 6	N 12.4		
Polyacrylonitrile	N 25		
PVC (100%)		Cl 56.7	
Medium Density Fibreboard (MDF)	N 3.7	Cl 0.62	
Wood (Pinus sylvestris)			
Plywood			
Polyisocyanurate (rigid PIR foam) – FR	N 6.2	Cl 2.53	
CMHR polyurethane flexible foam – FR ("blue foam")	N 8.2	Cl 2.5	
Acrylic/wool/polyester fabric "boucle"	N12.9	Cl 0.3	Br 0.5 S 0.94
Acrylic/wool/polyester fabric "boucle" back coated	N10.8	Cl 0.95	Br 6.1 S 0.76
Acrylic/cotton/polyester fabric "velour"	N 12.0		
Acrylic/cotton/polyester fabric "velour" back coated	N 7.3		Br 8.2

HER

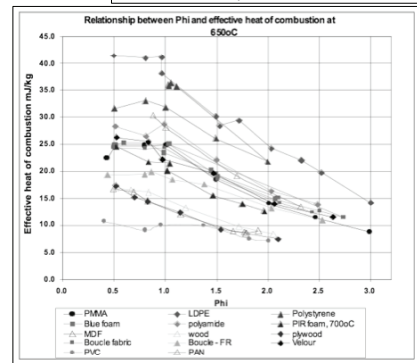
Nitrogen-containing materials tested in the tube furnace

Table 1: Characterisation of test materials

Material	Model 10, Rowland	Elemental composition (wt%)							
		C	H	N	O	S	Br	Cl	
Aliphatic polyurethane foam - FR	1.00	68.20	8.00	0.00	23.80	0.00	-	-	-
Polystyrene (PS) rigid foam	1.00	88.00	8.80	0.00	3.20	0.00	-	-	-
Polystyrene (PS) rigid foam	1.00	88.00	8.80	0.00	3.20	0.00	-	-	-
Acrylic/wool/polyester fabric "boucle"	1.00	68.20	8.00	0.00	23.80	0.00	-	-	-
Acrylic/wool/polyester fabric "boucle" back coated	1.00	68.20	8.00	0.00	23.80	0.00	-	-	-
Acrylic/cotton/polyester fabric "velour"	1.00	68.20	8.00	0.00	23.80	0.00	-	-	-
Acrylic/cotton/polyester fabric "velour" back coated	1.00	68.20	8.00	0.00	23.80	0.00	-	-	-
Acrylic/wool/polyester fabric "boucle"	1.00	68.20	8.00	0.00	23.80	0.00	-	-	-
Acrylic/wool/polyester fabric "boucle" back coated	1.00	68.20	8.00	0.00	23.80	0.00	-	-	-
Acrylic/cotton/polyester fabric "velour"	1.00	68.20	8.00	0.00	23.80	0.00	-	-	-
Acrylic/cotton/polyester fabric "velour" back coated	1.00	68.20	8.00	0.00	23.80	0.00	-	-	-
Acrylic/wool/polyester fabric "boucle"	1.00	68.20	8.00	0.00	23.80	0.00	-	-	-
Acrylic/wool/polyester fabric "boucle" back coated	1.00	68.20	8.00	0.00	23.80	0.00	-	-	-
Acrylic/cotton/polyester fabric "velour"	1.00	68.20	8.00	0.00	23.80	0.00	-	-	-
Acrylic/cotton/polyester fabric "velour" back coated	1.00	68.20	8.00	0.00	23.80	0.00	-	-	-

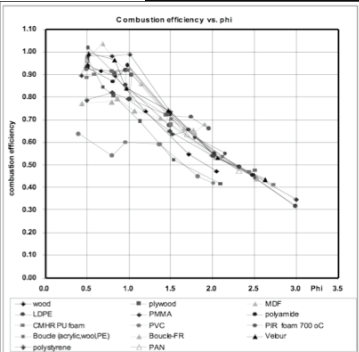
Elemental composition obtained: from elemental composition, analysis following bomb calorimetry or 900°C well-ventilated tube furnace runs

Yields versus equivalence ratios – BRE tube furnace



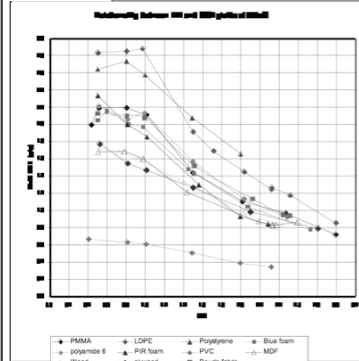
HER

Yields versus equivalence ratios – BRE tube furnace



- Most materials had combustion efficiencies of 0.90-1.0 (phi = 0.5), some decreasing to 80-90% (phi = 1) (aliphatics, cellulose n-containing)
- Aromatic (polystyrene) max 0.8
- FR-boucle fabric max 0.8
- PVC max 0.6
- Linear decreases above phi = 1 measured at 0.55 at phi = 2

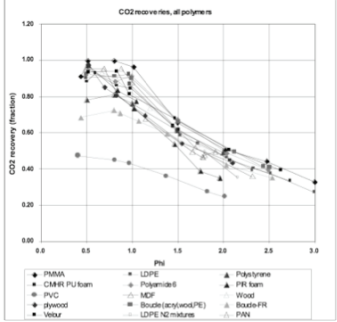
HER

CO₂ yields versus equivalence ratios – BRE tube furnace

- CO₂ yields similar to heat of combustion

HER

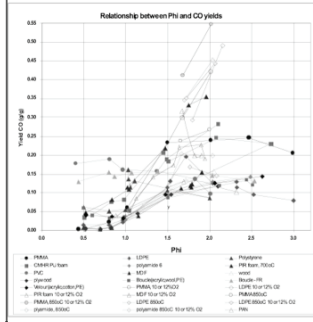
David Purser CO₂ yields versus equivalence ratios – BRE tube furnace



CO₂ recovery (fuel carbon mass loss as CO₂) similar to combustion efficiency

HER

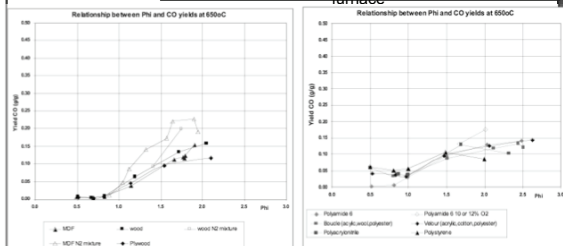
David Purser CO yields versus equivalence ratios – BRE tube furnace



- For materials leaving no residue phi was calculated from the stoichiometric oxygen demand of the whole material
- For char formers the char was considered as 100% carbon and unavailable as fuel for phi calculation
- Most materials have low CO yields (< 0.05 g/g mass loss) for phi < 1 (unless heavily FR)
- Phi > 1 sigmoid with a plateau ~ phi = 1.75-2.0
- Materials fell into groups:
 - High max yields up to 0.55 g/g : PMMA, Polyamide
 - Low max yields up to 0.14 g/g : cellulotics
 - Heavily halide FR materials (PVC, Boule-FR) – almost constant yield across range
- Effects of oxygen: for a given phi lower CO yields in air than in 10-12% O₂/N₂ mixtures
- Effects of temperature: at 850°C somewhat higher CO yields in some cases

HER

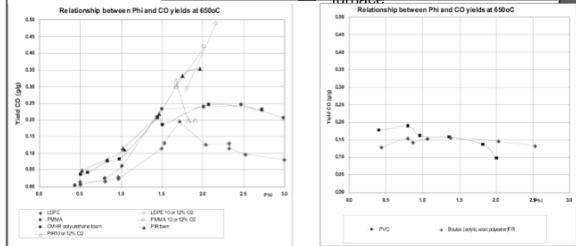
David Purser CO yields versus equivalence ratios – BRE tube furnace



- For some materials including cellulotics, aromatics and some other polymers the maximum CO yield was quite low, at 650°C, with a maximum around 0.15 g/g
- When 10-12% O₂ air / N₂ mixtures were used instead of 21% O₂ air, the maximum CO yields was increased to around 0.2 g/g

HER

David Purser CO yields versus equivalence ratios – BRE tube furnace



- For other materials including LDPE, PMMA, PIR and CMHR PU foam, the maximum CO yields were in air were higher at around 0.2-0.25 g/g mass loss.
- For LDPE and PMMA the maximum yields were increased to 0.42-0.5 g/g using air/N₂ mixtures, and even more at 850°C
- For PVC and the heavily FR-backed Boule fabric, the yield was more or less constant at ~ 0.15 g/g throughout the phi range

HER

David Purser Why do the CO yields vary between different materials?

CO *yields* may be useful to compare products but in order to understand the chemistry it is necessary to normalise the data for different materials

Differences in CO yield may be due to:

- Differences in carbon content – normalise yields in terms of carbon recovery as CO
- Char formation – express normalised yields in terms of total gas phase carbon
- Differences in reaction chemistry – can then be compared for the normalised data
- Attempts have also been made to derive algebraic expressions (using the Weibull distribution) for use with engineering calculations

HER

David Purser Normalized Yields

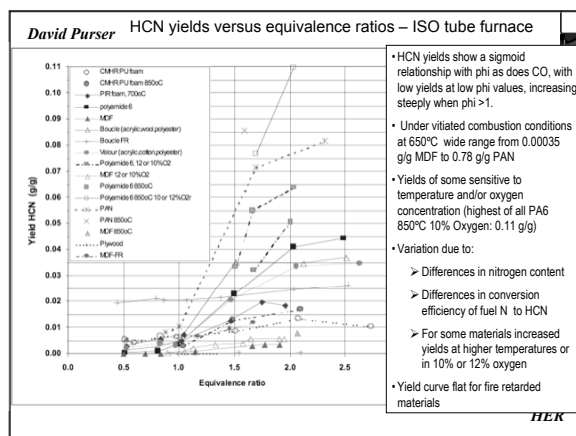
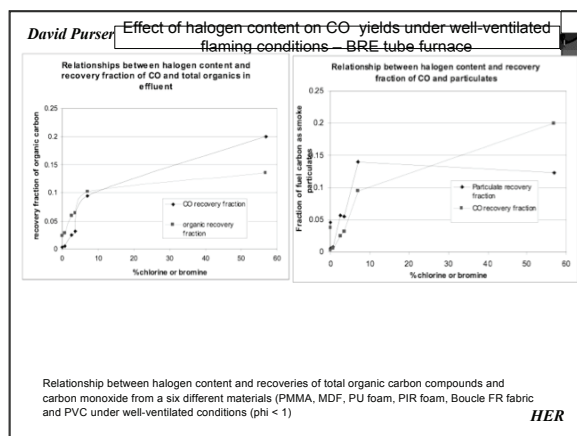
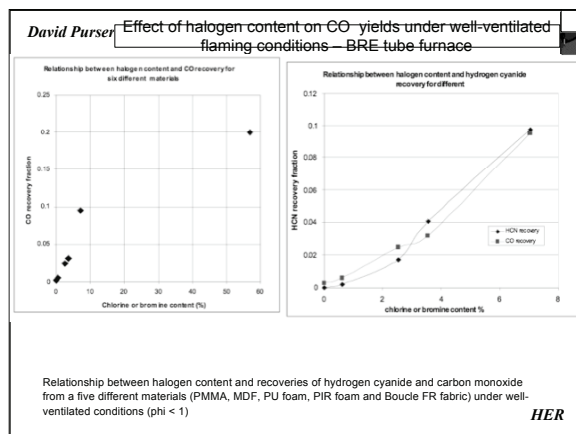
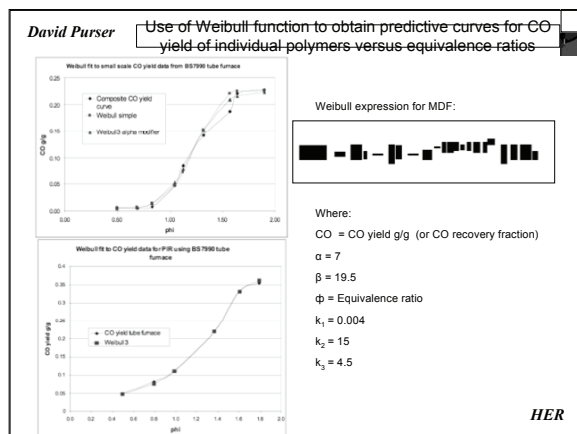
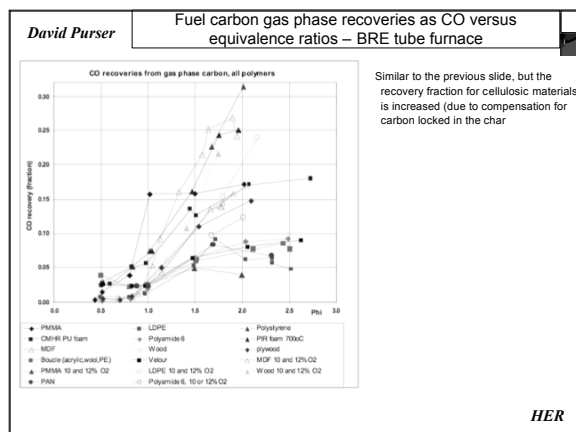
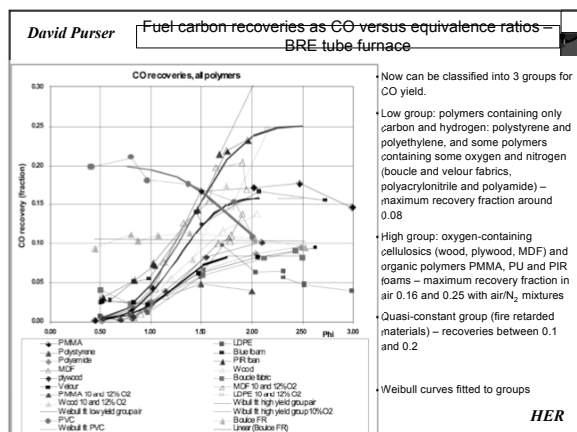
- Due to the variation in yields between different materials it is necessary to use different functions for the relationship between phi and yield for each individual material
- For engineering calculations it would be useful if the variation could be reduced to one or a few generic curves
- The main variable is nitrogen content – so plotted the normalised yields (recovery fractions (mass of fuel nitrogen released as nitrogen in HCN))
- Derived Weibull functions for generic sets of materials

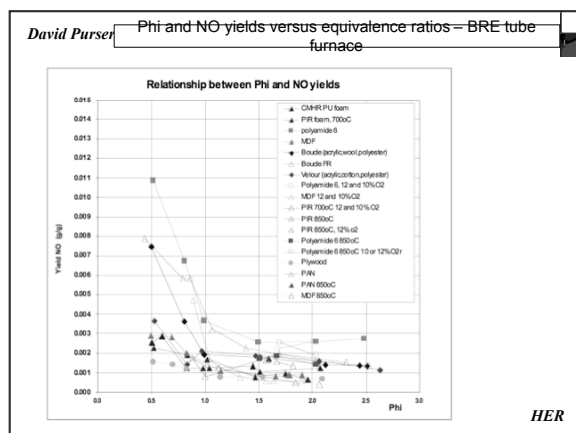
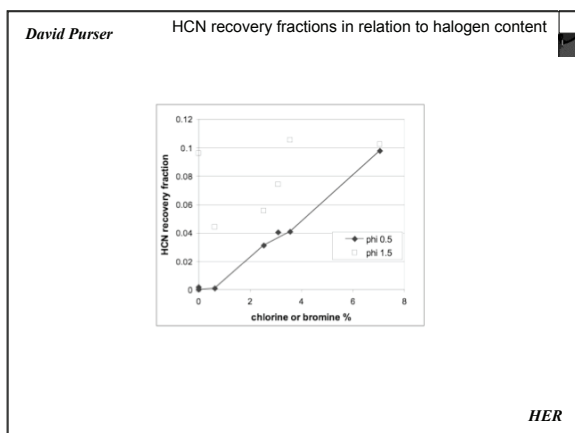
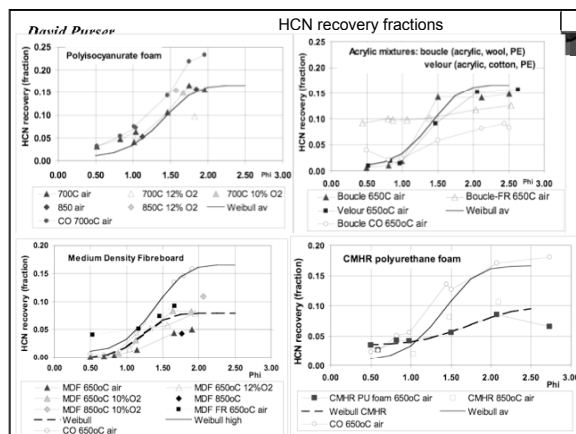
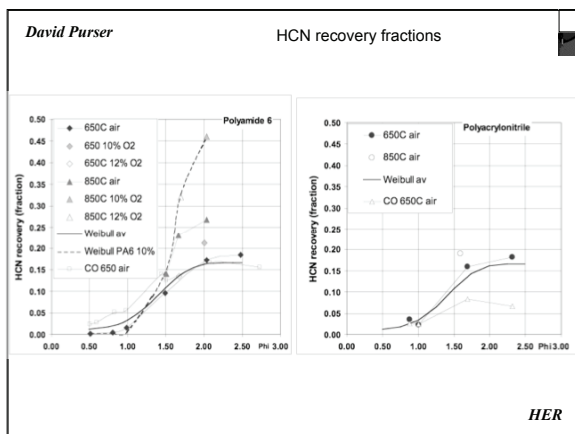
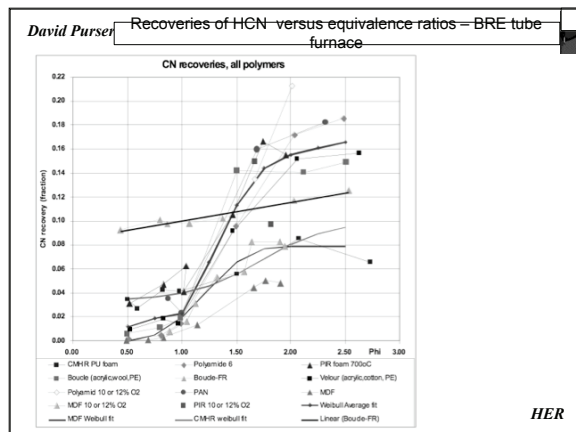
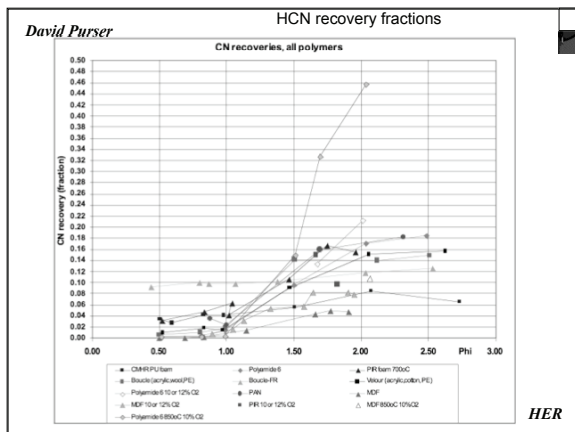


This function gives a good fit to the sigmoid relationships between phi and recovery fraction

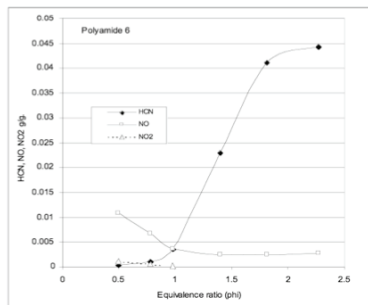
Material	Gamma	K	Gamma	K
LDPE recovery fraction	0.0	0.0	0.0	0.0
Polyamide 6 recovery fraction	0.0	0.0	0.0	0.0
High recovery fraction (wood polyamide)	0.0	0.0	0.0	0.0
LDPE recovery fraction	0.0	0.0	0.0	0.0
LDPE recovery fraction	0.0	0.0	0.0	0.0

HER





David Purser Yields of nitrogen-containing products from polyamide

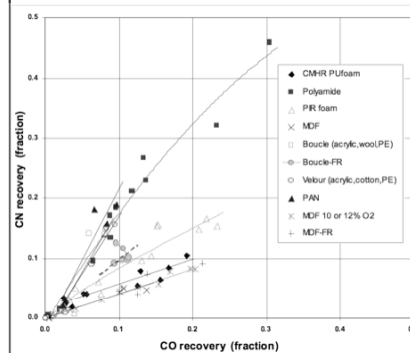


Relationship between equivalence ratio and yields of NO, NO₂ and HCN under flaming combustion conditions at 650°C in air measured in the steady state tube furnace

HER

David Purser

Conversion of fuel nitrogen to HCN compared to conversion of fuel carbon to CO – ISO 19700 tube furnace

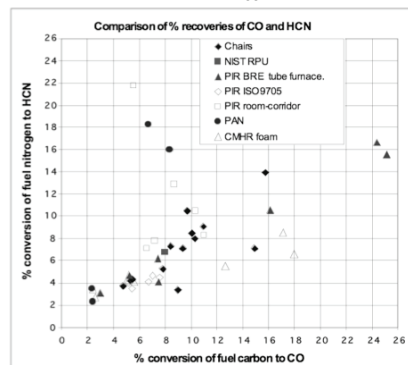


In fire tests and fire models, CO is measured but HCN seldom.

If CO and HCN recoveries are similar then HCN recovery can be estimated from CO recovery

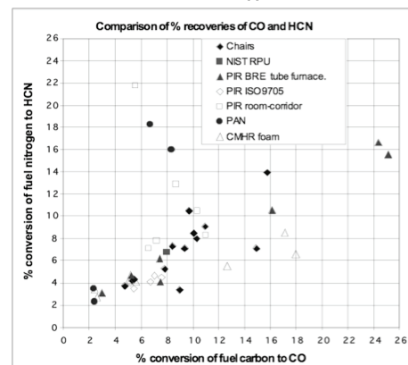
HER

David Purser Conversion of fuel nitrogen to HCN compared to conversion of fuel carbon to CO – ISO 19700 tube furnace and large-scale fires



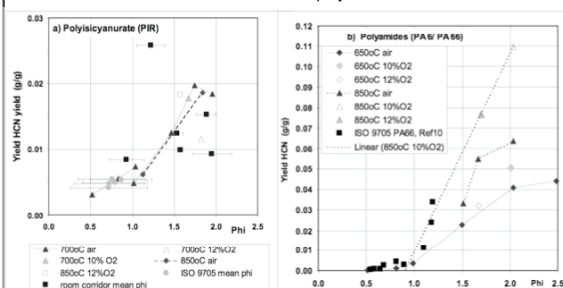
HER

David Purser Conversion of fuel nitrogen to HCN compared to conversion of fuel carbon to CO – ISO 19700 tube furnace and large-scale fires



HER

David Purser Comparison of HCN yields in large-scale fires and ISO 19700 tube furnace for polyisocyanurate foam and polyamide 6

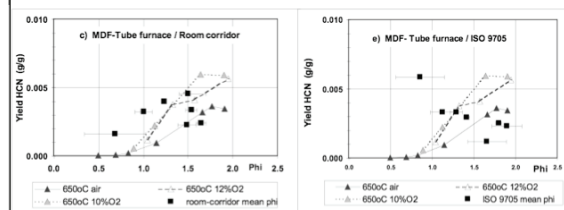


large-scale PA data from Blomqvist and Lönnermark using polyamide 66

HER

David Purser

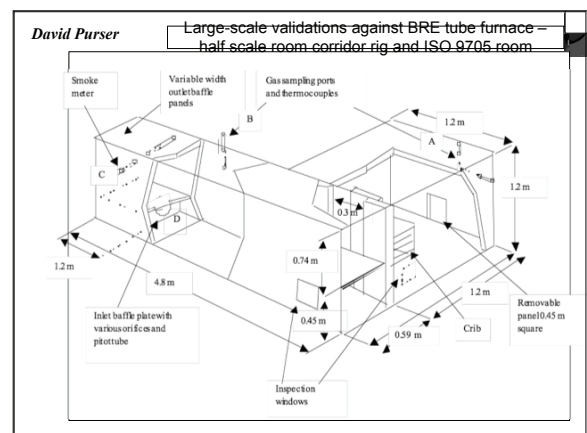
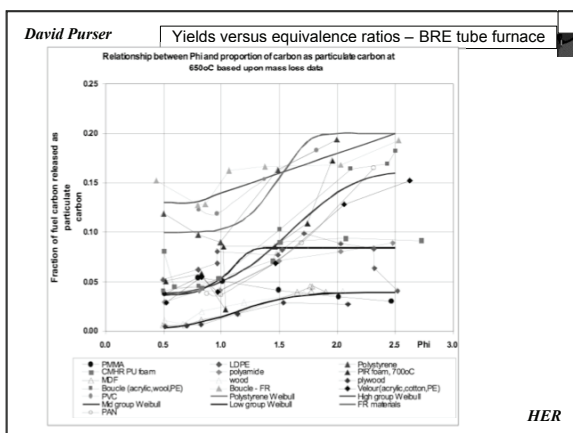
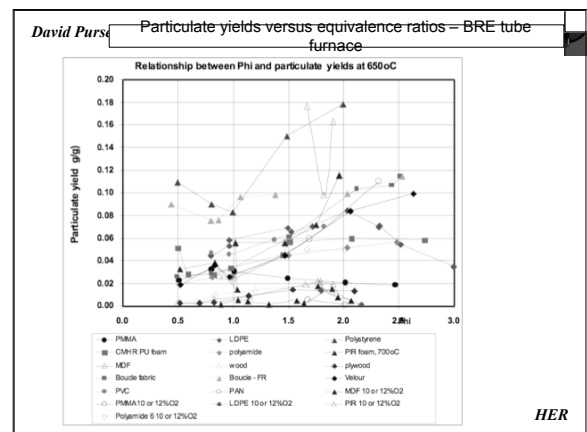
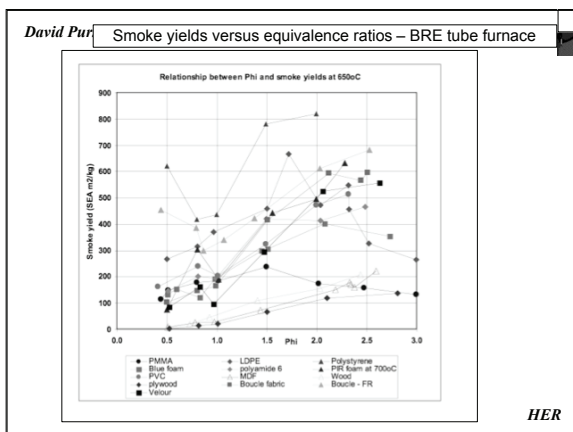
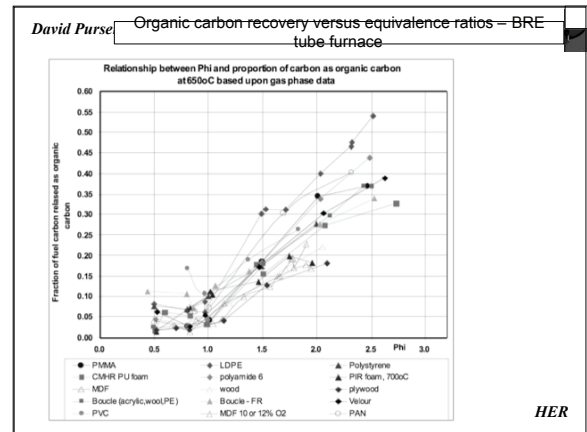
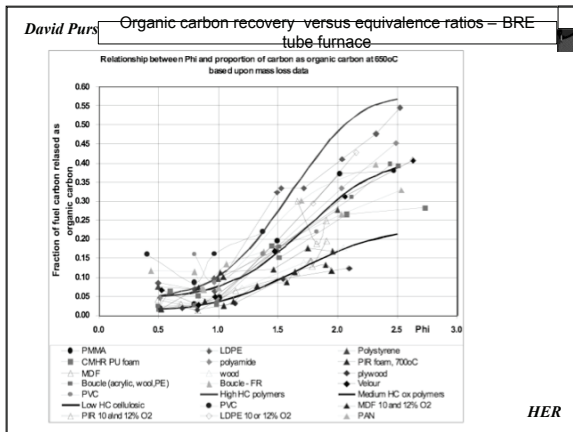
Comparison of HCN yields in large-scale fires and ISO 19700 tube furnace for polyisocyanurate foam and polyamide 6



Room corridor cribs only so fuel conditions well-defined

ISO 9705 room crib and wall linings so different combustion conditions in different parts of the room

HER



David Purser

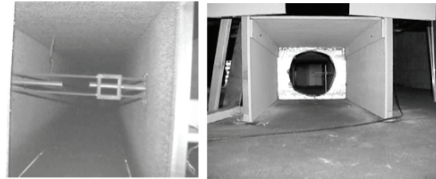
Large-scale validations against BRE tube furnace –
half scale room corridor rig and ISO 9705 room



HER

David Purser

Large-scale validations against BRE tube furnace –
half scale room corridor rig and ISO 9705 room

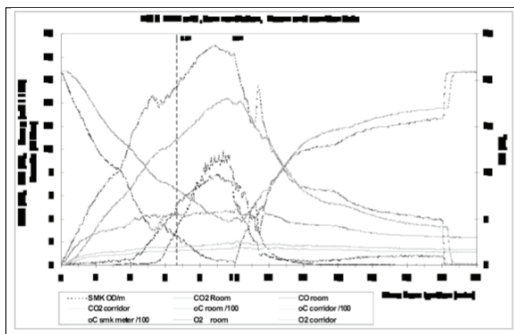


Upper section of corridor showing smoke meter and lower part
showing inlet baffle with pitot

HER

David Purser

Room-corridor experiment – MDF crib

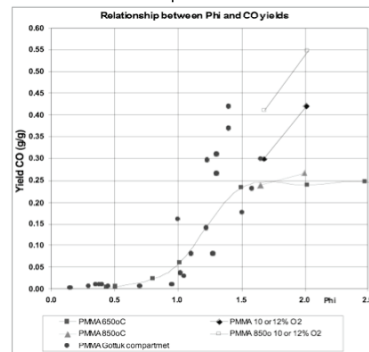


Room corridor test 3, MDF crib, 5.0 cm ventilation, Sample period: 6.67 – 9.67 min
Phi: min = 1.59, max = 1.70

HER

David Purser

Yields versus equivalence ratios – BRE tube furnace

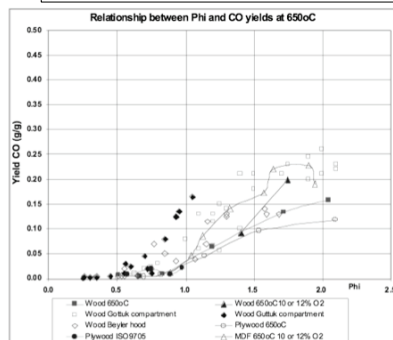


Comparison of CO yields (g/g) for PMMA obtained from the BRE tube-furnace apparatus in Gottuk's compartment fires

HER

David Purser

Yields versus equivalence ratios – BRE tube furnace

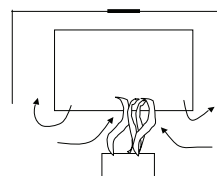


Comparison of CO yields (g/g) for wood and plywood obtained from the BRE tube-furnace apparatus, in Beyer hood experiments, Gottuk compartment fires and in the ISO 9705 room at BRE

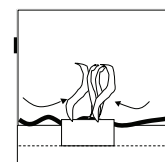
HER

David Purser

Yields versus equivalence ratios – BRE tube furnace



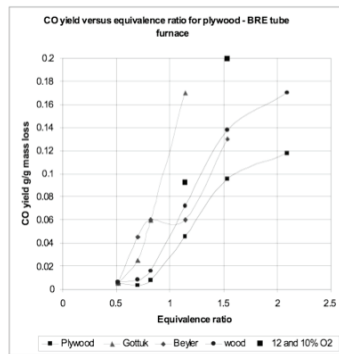
Beyer hood experiments,



Gottuk compartment fires

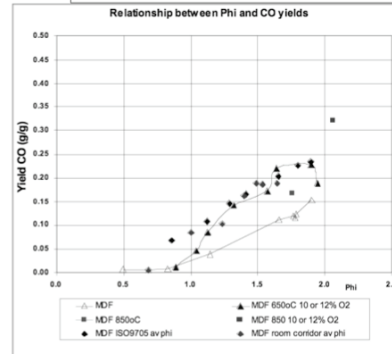
Similar results have since been obtained by Marlair using the Factory Mutual apparatus

HER

Effect of ϕ on CO yield

HER

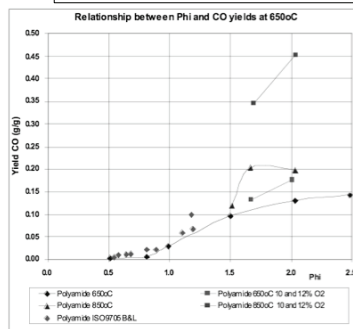
Yields versus equivalence ratios – BRE tube furnace



Comparison of CO yields (g/g) for MDF obtained from the BRE tube-furnace apparatus, in the ISO 9705 room and in the half-scale room corridor apparatus at BRE

HER

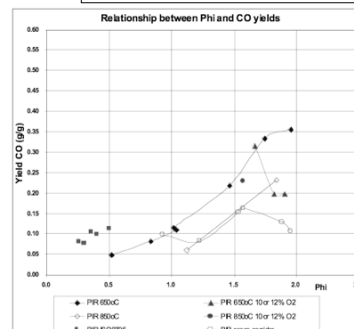
Yields versus equivalence ratios – BRE tube furnace



Comparison of CO yields (g/g) for Polyamide obtained from the BRE tube-furnace apparatus and in the ISO 9705 room by Blomqvist and Lonnemark

HER

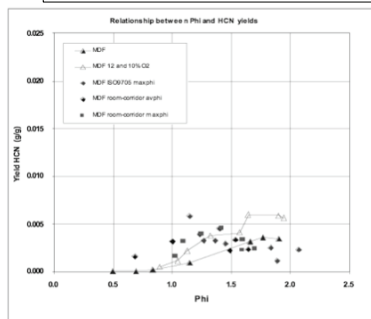
Yields versus equivalence ratios – BRE tube furnace



Comparison of CO yields (g/g) for PIR obtained from the BRE tube-furnace BRE apparatus, in the ISO 9705 room at BRE and in the and in the half-scale room corridor apparatus at BRE

HER

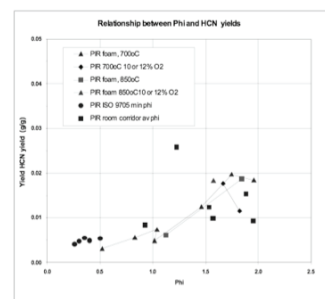
Yields versus equivalence ratios – BRE tube furnace



Comparison of HCN yields (g/g) for MDF obtained from the BRE tube-furnace apparatus, in the ISO 9705 room and in the half-scale room corridor apparatus at BRE

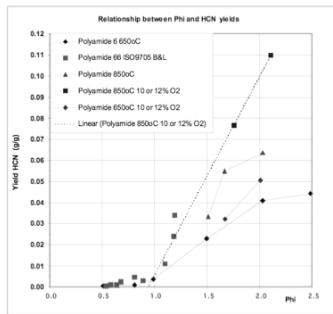
HER

Yields versus equivalence ratios – BRE tube furnace



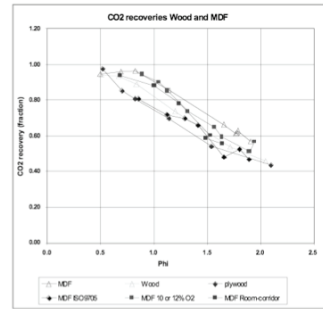
Comparison of HCN yields (g/g) for PIR obtained from the BRE tube-furnace apparatus, in the ISO 9705 room and in the half-scale room corridor apparatus at BRE

HER



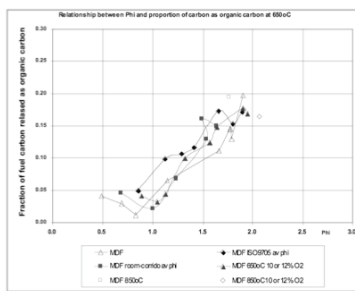
Comparison of HCN yields (g/g) for Polyamide obtained from the BRE tube-furnace apparatus in the ISO 9705 room by Blomqvist and Lonnemark

HER



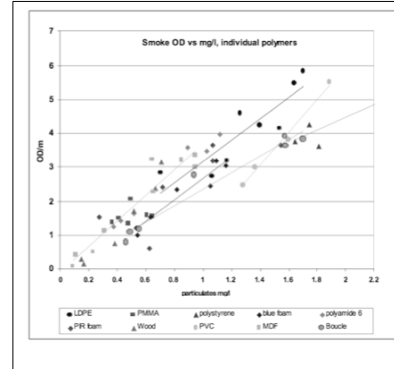
Comparison of CO₂ recovery fraction from PIR obtained from the BRE tube-furnace apparatus, in the ISO 9705 room and in the half-scale room corridor apparatus at BRE.

HER

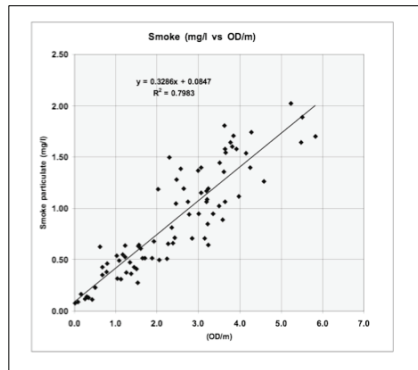


Comparison of carbon recovery as organic carbon for MDF obtained from the BRE tube-furnace apparatus, in the ISO 9705 room and in the half-scale room corridor apparatus at BRE.

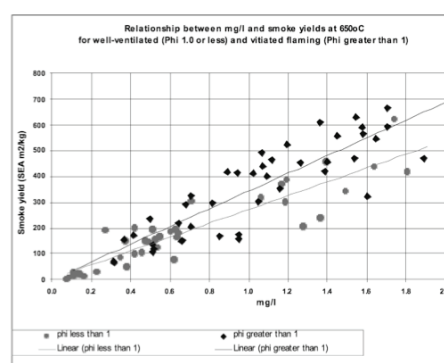
HER



HER



HER



HER

David Purser

Using ϕ and CO yield to model hazard in a full scale fire

- A full scale fire is estimated as a medium growth rate t^2 fire
- The fire is in an enclosed room (2.44 x 3 x 2.44 m) with a 1 m x 0.3 m vent at floor level
- The fire growth heat release rate and upper zone filling rate are modelled using C-fast
- The mass loss rate is calculated from the heat release rate
- The CO yield is taken from the tube furnace experiments as:

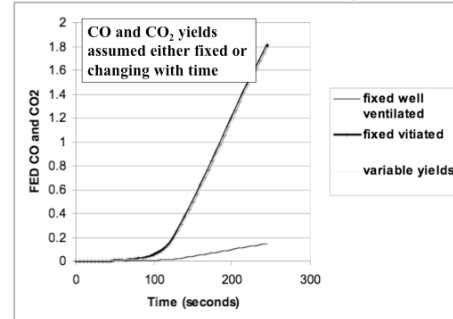
well ventilated 0.007 CO 2.54 g/g fuel mass loss
vitiated 0.149 CO 1.57

The FED for incapacitation from CO and CO₂ is calculated

HER

David Purser

Using ϕ and CO yield to model hazard in a full scale fire



HER

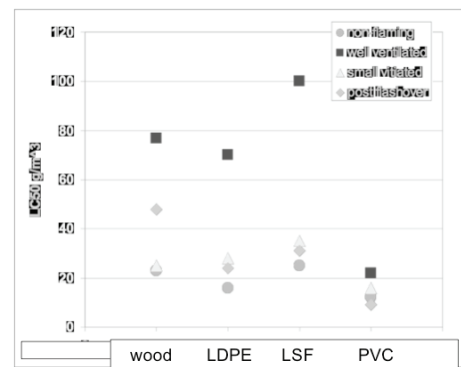
David Purser

Using ϕ and CO yield to model hazard in a full scale fire

- This fire was assumed to be a polyethylene fire, what difference (with respect to CO and CO₂ hazard) would it have made if it had been a PVC fire?
- Under well ventilated conditions, the yield of CO from PVC was 0.166g/g which is 24 x the yield from PE, the toxic potency is therefore greater
- Under vitiated conditions the yield of CO from PE was 0.149 g/g which is close to that of PE
- Since the modelling shows only the vitiated yields determine the time to incapacitation, it would have made no difference if PVC had been used and grew at the same rate. In practice the PVC fire would grow more slowly, giving a longer time to asphyxiation.

David Purser

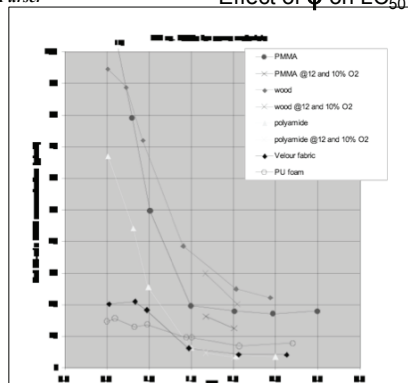
Summary of toxic potency estimation



HER

David Purser

Effect of ϕ on LC₅₀



HER

David Purser

Conclusions

- The BS7990 tube furnace provides a versatile method for measuring toxic product yields over the full range of fire conditions
- For flaming fires the yields from individual materials are very dependent upon fuel/air ratios (ϕ) and to a lesser extent on oxygen concentrations and temperature
- Fire retardants, especially halides also have a large effect on toxic product yields
- For flaming fires, yields expressed as a function of equivalence ratio appear to be apparatus independent and correlate well with those obtained from other test methods and large-scale fires
- Functions can be derived for the prediction of toxic product yields against ϕ for application to FSE calculations

HER